

An Introduction to Atmospheric Radiation

SECOND EDITION

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The annual global insolation is proportional to $(1 + e^2/2)$, but is independent of the declination of the sun δ and the true anomaly ν .

2.3 Solar Spectrum and Solar Constant Determination

2.3.1 Solar Spectrum

The solar spectrum covers wavelengths ranging from gamma rays to radio waves, as shown in Fig. 1.1. Because of the nonquantized electronic transitions, most solar energy is carried by the continuum, i.e., radiation is continuous rather than selective. The single most important contributor is hydrogen, both in its neutral state and as negative ions. A radiation transition from one level to another is characterized by an absorption or an emission line whose frequency is governed by Planck's relation. However, in the ionization process the atom (or molecules) may absorb more than the minimum energy required to remove the electron. This additional energy may be thought of as supplying kinetic energy to the freed electron and is not quantized. As a consequence, absorption is not selective but rather continuous. The ionization continuum occurs on the high-frequency (shorter wavelength) side of the ionization frequency. Neutral hydrogen has ionization continua associated with lines, some of which were defined in Fig. 1.9. Metallic atoms also contribute to the continuum in the ultraviolet spectrum. The continuum absorption in the visible and infrared spectrum, however, is produced by negative hydrogen ions.

Electromagnetic radiation emerging from within the sun is continuously emitted and absorbed by atoms. As shown in Fig. 2.2, the radiative temperature first drops off to a minimum value of about 4500 K just above the photosphere, and then levels off and slowly rises in the chromosphere, followed by a rapid rise in the transition region to several million degrees in the corona. At each temperature, probabilities of the electronic transition exist that any atom will achieve a particular excited state, leading to the formation of absorption lines at different levels in the solar atmosphere. The core of a line forms at the temperature where the maximum transition probabilities of an electron moving from one orbital level to another occur (see Fig. 1.8). The wings of a line form at different temperature levels because of the required transition probabilities. Each absorption line has a preferred formation region in the solar atmosphere. Those lines that absorb very little radiation are known as weak lines, which can form in narrow layers of the solar atmosphere. Some of the absorption lines in the solar atmosphere were displayed in Fig. 2.2.

In view of the preceding discussion, the solar spectrum consists of a continuous emission with a superimposed line structure. The visible and infrared spectrum of the photosphere shows absorption lines, known as the *Fraunhofer spectrum*. The strongest of these lines are produced by H, Mg, Fe, Ca, and Si, as well as singly ionized Ca and Mg. Most of the lines shorter than 1850 Å produced from the photosphere exhibit in emission. Light from the chromosphere and the corona has emission lines at all observed wavelengths.

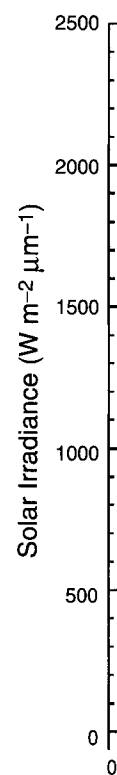


Figure 2.9 Solar results presented in the 5800 K accounting for

Figure 2.9 shows the solar spectrum averaged over a 5 μm band, based on the data of (Kopp and 1995). Although the spectral solar constant is proposed solar constant of 1361 W m⁻², the 50 cm⁻¹ spectral irradiance is produced by the absorption of the solar radiation. As can be seen, particularly in the visible and infrared, an emitting temperature of the sun and the earth's atmosphere spectrum characterizes the solar radiation. In applications, it is used in radiative transfer calculations.

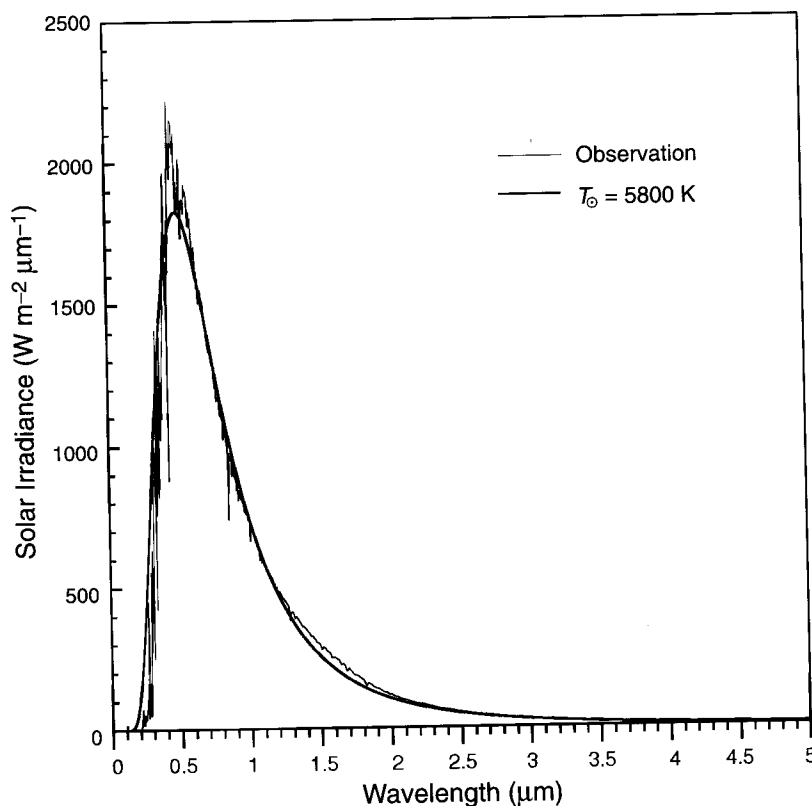


Figure 2.9 Solar irradiance for a 50 cm^{-1} spectral interval at the top of the atmosphere based on the results presented in the MODTRAN 3.7 program. Also shown is the Planck flux with a temperature of 5800 K accounting for the mean distance between the earth and the sun.

Figure 2.9 shows the spectral solar irradiance observation at the top of the atmosphere averaged over a 50 cm^{-1} spectral interval as a function of wavelength up to $5 \mu\text{m}$, based on the results presented in the MODTRAN 3.7 program (Anderson *et al.*, 1995). Although the total solar irradiance derived from this program is 1373 W m^{-2} , the spectral solar irradiance curve presented here is scaled with respect to the recently proposed solar constant of 1366 W m^{-2} (see Section 2.3.3 for further discussion). A 50 cm^{-1} spectral average has been performed to smooth out the rapid fluctuations produced by the absorption/emission line structure. However, some variabilities can still be seen, particularly in the ultraviolet spectrum. Also shown is the Planck curve with an emitting temperature of 5800 K, taking into account the mean distance between the sun and the earth. This temperature appears to fit closely with the visible and infrared spectrum characteristic of radiation emitted from the photosphere. For atmospheric applications, it is critically important to have reliable spectral solar irradiances for use in radiative transfer models. Table 2.3 gives tabulated data from 0.2 to $5 \mu\text{m}$ with

Table 2.3

Distribution of Solar Spectral Irradiance S_λ from 0.2 to 100 μm in Terms of the Accumulated Energy and Percentage Based on the Values Listed in the MODTRAN 3.7 Program^a

λ (μm)	S_λ ($\text{W m}^{-2} \mu\text{m}^{-1}$)	$S_{0-\lambda}$ (W m^{-2})	$S_{0-\lambda}$ (%)	λ (μm)	S_λ ($\text{W m}^{-2} \mu\text{m}^{-1}$)	$S_{0-\lambda}$ (W m^{-2})	$S_{0-\lambda}$ (%)
0.20	2.0832E+01	2.08317E+00	0.15250	3.8	1.0564E+01	1.35239E+03	99.00390
0.30	5.4765E+02	5.68479E+01	4.16163	3.9	9.6162E+00	1.35335E+03	99.07430
0.40	1.4042E+03	1.97272E+02	14.44155	4.0	8.6980E+00	1.35422E+03	99.13797
0.50	1.9619E+03	3.93464E+02	28.80410	4.1	7.9180E+00	1.35502E+03	99.19593
0.60	1.7632E+03	5.69780E+02	41.71153	4.2	7.2072E+00	1.35574E+03	99.24870
0.70	1.4300E+03	7.12778E+02	52.17994	4.3	6.5062E+00	1.35639E+03	99.29633
0.80	1.1257E+03	8.25347E+02	60.42075	4.4	5.7954E+00	1.35697E+03	99.33875
0.90	8.8835E+02	9.14182E+02	66.92404	4.5	5.2622E+00	1.35749E+03	99.37727
1.00	7.2943E+02	9.87125E+02	72.26392	4.6	4.8180E+00	1.35798E+03	99.41255
1.10	5.8743E+02	1.04587E+03	76.56425	4.7	4.4724E+00	1.35842E+03	99.44529
1.20	4.8921E+02	1.09479E+03	80.14558	4.8	4.1565E+00	1.35884E+03	99.47573
1.30	4.0851E+02	1.13564E+03	83.13614	4.9	3.8504E+00	1.35922E+03	99.50391
1.40	3.4450E+02	1.17009E+03	85.65813	5.0	3.5740E+00	1.35958E+03	99.53008
1.50	2.9066E+02	1.19916E+03	87.78592	6.0	1.8385E+00	1.36303E+03	99.78240
1.60	2.4644E+02	1.22380E+03	89.58999	7.0	1.0108E+00	1.36404E+03	99.85639
1.70	2.0453E+02	1.24425E+03	91.08726	8.0	5.9672E-01	1.36464E+03	99.90007
1.80	1.6829E+02	1.26108E+03	92.31927	9.0	3.7458E-01	1.36501E+03	99.92751
1.90	1.3725E+02	1.27481E+03	93.32404	10.0	2.4702E-01	1.36526E+03	99.94559
2.00	1.1624E+02	1.28643E+03	94.17501	11.0	1.6932E-01	1.36543E+03	99.95798
2.10	9.7416E+01	1.29617E+03	94.88816	12.0	1.2005E-01	1.36555E+03	99.96677
2.20	8.2132E+01	1.30439E+03	95.48942	13.0	8.7276E-02	1.36563E+03	99.97315
2.30	6.9594E+01	1.31134E+03	95.99889	14.0	6.5062E-02	1.36570E+03	99.97792
2.40	5.9198E+01	1.31726E+03	96.43226	15.0	4.9463E-02	1.36575E+03	99.98154
2.50	5.1023E+01	1.32237E+03	96.80577	16.0	3.8307E-02	1.36579E+03	99.98434
2.60	4.4280E+01	1.32679E+03	97.12994	17.0	3.0112E-02	1.36582E+03	99.98655
2.70	3.8672E+01	1.33066E+03	97.41305	18.0	2.3991E-02	1.36584E+03	99.98831
2.80	3.3815E+01	1.33404E+03	97.66058	19.0	1.9351E-02	1.36586E+03	99.98973
2.90	2.9589E+01	1.33700E+03	97.87720	20.0	1.5797E-02	1.36588E+03	99.99088
3.00	2.6133E+01	1.33962E+03	98.06850	30.0	3.4388E-03	1.36598E+03	99.99860
3.10	2.3093E+01	1.34193E+03	98.23756	40.0	1.0465E-03	1.36599E+03	99.99937
3.20	2.0476E+01	1.34397E+03	98.38746	50.0	4.2098E-04	1.36600E+03	99.99968
3.30	1.8186E+01	1.34579E+03	98.52059	60.0	2.0151E-04	1.36600E+03	99.99983
3.40	1.6191E+01	1.34741E+03	98.63913	70.0	1.0860E-04	1.36600E+03	99.99991
3.50	1.4562E+01	1.34887E+03	98.74574	80.0	6.3779E-05	1.36600E+03	99.99995
3.60	1.3032E+01	1.35017E+03	98.84114	90.0	3.9985E-05	1.36600E+03	99.99998
3.70	1.1670E+01	1.35134E+03	98.92657	100.0	2.6459E-05	1.36600E+03	100.0000

^aThe solar constant is taken to be 1366 W m^{-2} .

a $0.1\text{-}\mu\text{m}$ spectral interval. From 5 to 100 μm , solar irradiance accounts for about 6 W m^{-2} . Based on these values, about 50% of the total solar irradiance lies in wavelengths longer than the visible, about 40% in the visible region, and about 10% in wavelengths shorter than the visible. Note that from 3.5 to 5 μm , the emitted thermal infrared radiation from the earth and the atmosphere system becomes significant.

According to solar flux observations, the ultraviolet region ($<0.4 \mu\text{m}$) of the solar spectrum deviates greatly from the visible and infrared regions in terms of

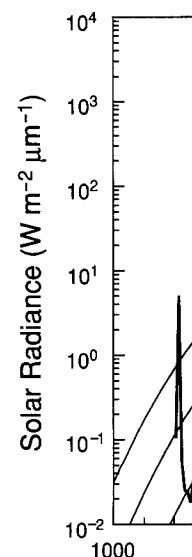


Figure 2.10 Observed solar spectrum from Brasseur and Simonovic, 1981, 4500 K to 6000 K.

the equivalent blackbody temperature of the observed solar spectrum. The temperatures of 4500, 5000, and 6000 K correspond to blackbody temperatures. The curve drops to a minimum level around 1.2 μm , then rises sharply to a maximum at 1216 \AA associated with the Lyman series of hydrogen atoms. The curve then drops to a relatively small amount around 1.6 μm and atomic oxygen. The curve then rises to represent the prime

2.3.2 Determination of the Solar Spectrum

For historical reasons, the determination of the solar spectrum for the purpose of determining the solar constant is the *pyrheliometer*. This instrument is used to measure the solar constant, utilizing a suitable instrument, measure

Terms of the Accumulated Energy and
DTRAN 3.7 Program^a

S_{λ} ($\text{W m}^{-2} \mu\text{m}^{-1}$)	$S_{0-\lambda}$ (W m^{-2})	$S_{0-\lambda}$ (%)
0564E+01	1.35239E+03	99.00390
5162E+00	1.35335E+03	99.07430
6980E+00	1.35422E+03	99.13797
9180E+00	1.35502E+03	99.19593
2072E+00	1.35574E+03	99.24870
5062E+00	1.35639E+03	99.29633
7954E+00	1.35697E+03	99.33875
2622E+00	1.35749E+03	99.37727
8180E+00	1.35798E+03	99.41255
7724E+00	1.35842E+03	99.44529
1565E+00	1.35884E+03	99.47573
3504E+00	1.35922E+03	99.50391
5740E+00	1.35958E+03	99.53008
8385E+00	1.36303E+03	99.78240
9108E+00	1.36404E+03	99.85639
9672E-01	1.36464E+03	99.90007
7458E-01	1.36501E+03	99.92751
4702E-01	1.36526E+03	99.94559
6932E-01	1.36543E+03	99.95798
2005E-01	1.36555E+03	99.96677
7276E-02	1.36563E+03	99.97315
5062E-02	1.36570E+03	99.97792
9463E-02	1.36575E+03	99.98154
8307E-02	1.36579E+03	99.98434
0112E-02	1.36582E+03	99.98655
9991E-02	1.36584E+03	99.98831
9351E-02	1.36586E+03	99.98973
5797E-02	1.36588E+03	99.99088
4388E-03	1.36598E+03	99.99860
0465E-03	1.36599E+03	99.99937
2098E-04	1.36600E+03	99.99968
0151E-04	1.36600E+03	99.99983
0860E-04	1.36600E+03	99.99991
3779E-05	1.36600E+03	99.99995
9985E-05	1.36600E+03	99.99998
6459E-05	1.36600E+03	100.0000

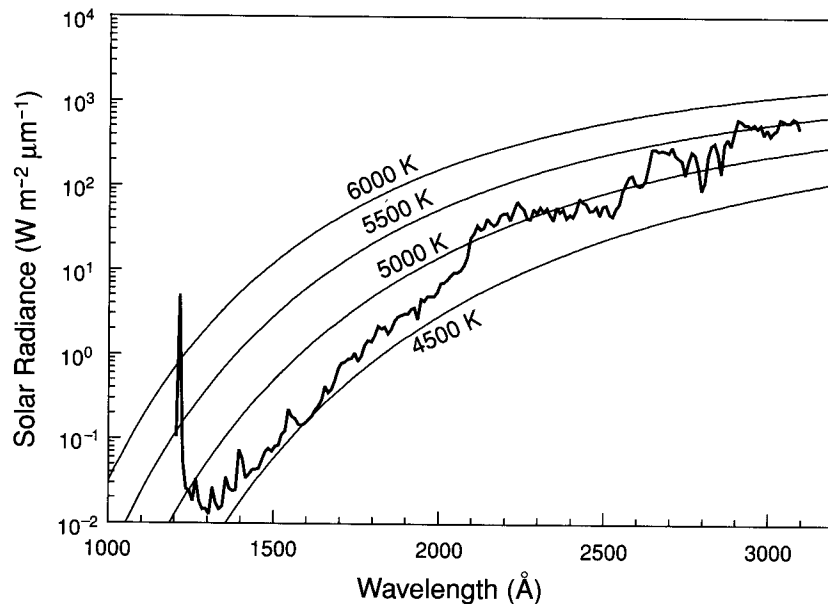


Figure 2.10 Observed irradiance outside the earth's atmosphere in the ultraviolet region (data taken from Brasseur and Simon, 1981) and comparison with the Planck curves for temperatures ranging from 4500 K to 6000 K.

the equivalent blackbody temperature of the sun. Figure 2.10 illustrates a detailed observed solar spectrum from about 1000 to 3000 Å, along with blackbody temperatures of 4500, 5000, 5500, and 6000 K. In the interval 2100–3000 Å, the equivalent blackbody temperature of the sun lies somewhat above 5000 K. It falls gradually to a minimum level of about 4700 K at about 1400 Å. From there toward shorter wavelengths, a larger amount of energy flux is observed at the Lyman α emission line of 1216 Å associated with the transition of the first excited and ground states of hydrogen atoms. The ultraviolet portion of the solar spectrum below 3000 Å contains a relatively small amount of energy. However, because the ozone and the molecular and atomic oxygen and nitrogen in the upper atmosphere absorb all this energy, it represents the prime source of the energy in the atmosphere above 10 km.

2.3.2 Determination of the Solar Constant: Ground-Based Method

For historical reasons, we shall first introduce the ground-based method for the determination of the solar constant. Ground-based observations of solar irradiance for the purpose of determining the solar constant require three primary instruments. These are the *pyrheliometer*, the *pyranometer*, and the *spectrobolometer*. The *pyrheliometer* is used to measure the direct, plus some diffuse, solar radiation, while the *pyranometer*, utilizing a suitable shield to block the direct solar radiation from striking the instrument, measures only the diffuse solar radiation for arriving at a *pyrheliometer*.

correction. The amount of direct sunlight can then be calculated by subtracting the flux density measured by the pyranometer from that measured by the pyrliometer. The spectrobolometer is a combination of a spectrograph and a coelostat. A coelostat is a mirror that follows the sun and focuses its rays continuously on the entrance slit of the spectrograph, which disperses the solar radiation into different wavelengths by means of a prism or diffraction grating. In the Smithsonian solar constant measurements, about 40 standard wavelengths between 0.34 and 2.5 μm are measured nearly simultaneously from the record of the spectrograph. The instrument for these measurements is called a *bologram*. There are two techniques of measuring the solar constant from the ground-based radiometer, called the *long* and *short* methods of the Smithsonian Institution. The long method is more fundamental and establishes the basis for the short method. The long method uses the Beer-Bouguer-Lambert law and is introduced in the following.

Consider an atmosphere consisting of plane-parallel layers. At a given position of the sun, which is denoted by the solar zenith angle θ_0 , the effective path length of the air mass is $u \sec \theta_0$, where

$$u = \int_{z_1}^{z_\infty} \rho dz. \quad (2.3.1)$$

In this equation z_1 is the height of the station and z_∞ denotes the top of the atmosphere. On the basis of the Beer-Bouguer-Lambert law, the irradiance F of the direct solar radiation of wavelength λ observed at the surface level is given by

$$F_\lambda = F_{\lambda 0} \exp(-k_\lambda u \sec \theta_0) = F_{\lambda 0} T_\lambda^m, \quad (2.3.2)$$

where $F_{\lambda 0}$ is the monochromatic solar irradiance at the top of the atmosphere, k_λ denotes the monochromatic mass extinction cross section, T_λ is the monochromatic transmissivity defined in Eq. (1.4.10), and $m (= \sec \theta_0)$ represents the ratio of the air mass between the sun and the observer to the air mass with respect to the local zenith distance. Upon taking the logarithm, we find

$$\ln F_\lambda = \ln F_{\lambda 0} + m \ln T_\lambda. \quad (2.3.3)$$

Observations of F_λ may be made for several zenith angles during a single day. If the atmospheric properties do not change during the observation period, then the transmissivity T_λ is constant. A plot of $\ln F_\lambda$ versus m shown in Fig. 2.11 may be extrapolated to the zero point, which represents the top of the atmosphere ($m = 0$). This is referred to as the *Langley plot*. If observations of the monochromatic irradiance are carried out for wavelengths covering the entire solar spectrum, then we have

$$F_\odot = \int_0^\infty F_{\lambda 0} d\lambda \approx \sum_{i=1}^N F_{\lambda_i 0} \Delta \lambda_i, \quad (2.3.4)$$

where N is the total number of the monochromatic irradiances measured. The irradiance F_\odot corresponds to the actual distance between the earth and the sun, r . By using

Figure 2.11
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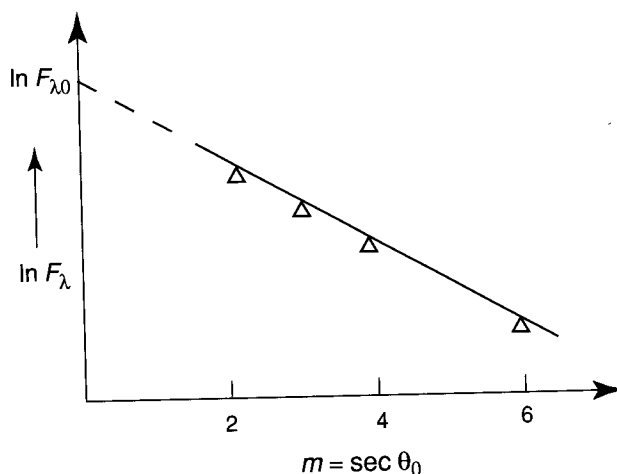


Figure 2.11 Hypothetical observed monochromatic solar irradiances F_λ as a function of the effective path length from which the solar irradiance at the top of the atmosphere can be graphically determined, referred to as the Langley plot.

(2.3.1)

the mean distance $r_0 = a$, the solar constant is defined by

$$S = F_\odot(r/a)^2. \quad (2.3.5)$$

The foregoing outlines the theoretical procedures of the Smithsonian long method for the determination of the solar constant. However, the atmosphere is essentially opaque for wavelengths shorter and longer than about $0.34 \mu\text{m}$ and $2.5 \mu\text{m}$, respectively. Consequently, flux density observations cannot be made in these regions. Therefore, empirical corrections are needed for the omitted ranges, which account for about 8% of the solar flux.

There are other sources of error inherent in the Smithsonian long method caused by (1) empirical corrections for the absorption of ultraviolet by ozone, and the absorption of infrared by water vapor and carbon dioxide in the wings of the solar spectrum; (2) an unknown amount of diffuse radiation entering the aperture of the observing instrument; (3) variations of k_λ and the possible effects of aerosols during a series of measurements; and (4) measurement errors. Therefore, in spite of careful evaluation and observation, a certain amount of error is inevitable.

Employing the Smithsonian long method, each determination requires about 2 to 3 hours of observation time, plus twice that much time for the data reduction. In addition, there is no assurance that atmospheric properties and solar conditions will remain unchanged during the observation period. Because of this uncertainty and the burdensome, time-consuming work involved, a short method was devised to determine the solar constant.

In the short method, the diffuse component of solar radiation (the sky brightness) is measured for a given locality over a long period of time, so that a mean diffuse intensity

(2.3.3)

(2.3.4)

can be determined. Thus, a pyranometer reading of diffuse solar radiation will differ from the mean by an amount ε , called the *pyranometer excess*. In reference to Section 1.1.4 and Fig. 3.9, the attenuation of solar radiation on a clear day is due to scattering by molecules and aerosol particles, and absorption by various gases, primarily water vapor. If total precipitable water is given by w , an empirical relationship between the attenuation of direct solar irradiance and scattering and absorption effects may be expressed in the form $F_\lambda = w + q_\lambda \varepsilon$, where q_λ is a constant empirically determined for each wavelength for a given locality. With q_λ known, the spectral value of the solar irradiance can be found from the observed precipitable water and a pyranometer reading.

On the basis of a long series of previous observations of F_λ , m , and T_λ at a given location where the solar constant measurement has been made, a graph of F_λ versus air mass m can be constructed for a set value of T_λ . Thus, for a particular measurement of F_λ with a known air mass m , the corresponding transmissivity T_λ can be found from the graph. Once T_λ has been determined, solar irradiance at the top of the atmosphere $F_{0\lambda}$ can be evaluated through Eq. (2.3.2). After this point, evaluation of the solar constant proceeds in the same manner as in the long method. In the short method, the required measurements include a bologram of the sun, an observation of sky brightness by the pyranometer, and air mass determined by the position of the sun from a theodolite. These three measurements take only about 10 to 15 minutes. From the thousands of observations at various locations around the world during a period of more than half a century, the best value of the solar constant determined by the Smithsonian methods is 1353 W m^{-2} .

The presence of aerosols in the atmosphere imposes limitations on the accuracy of ground-based radiometric measurements of the solar constant (see, e.g., Reagan *et al.*, 1986). To minimize atmospheric effects, a number of measurements have also been made in the upper atmosphere and outer space. These have included observations made from balloons floating in the 27- to 35-km altitude range, jet aircraft at about 12 km, the X-15 rocket aircraft at 82 km, and the Mars Mariner VI and VII spacecrafts entirely outside the atmosphere. The solar constant derived from these experiments varies. Based on a series of measurements from high-altitude platforms, a standard solar constant of $1353 (\pm 21) \text{ W m}^{-2}$ was issued in 1976 by the National Aeronautics and Space Administration (Thekaekara, 1976).

2.3.3 Satellite Measurements of the Solar Constant

Measurements of incoming solar irradiance have been routinely made from satellite platforms since the mid-1970s. However, the high-accuracy, high-stability satellite-borne radiometer was only developed and incorporated in the Nimbus 7 satellite in 1978. This radiometer was an electrically calibrated cavity radiometer. The basic concept of blackbody cavity radiation was shown in Fig. 1.6. Radiometers of this design for use in satellites had a black painted cavity that absorbed nearly all the solar radiation impinging on it. The absorbed radiation raised the temperature of the cavity so that a radiant power could be measured corresponding to the increase in temperature. Such a cavity can also be heated by an electrical element in a manner equivalent to the incident sunlight. Because the input electrical power can be measured

accurately, the temperature referred to as the *self*

Solar constant data made by self-calibrating include the Nimbus Maximum Mission (S in 1980; the Earth Radiation Budget Satellite and the ACRIM II mission (UARS, 1991). Figure these satellites from daily data. The absolute

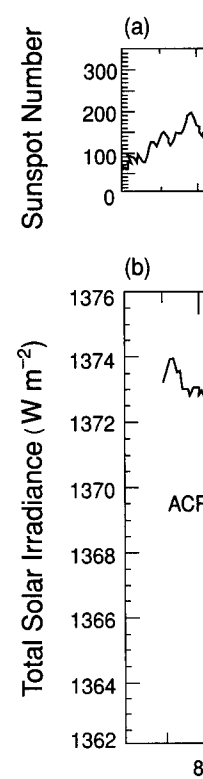


Figure 2.12 Solar ac (b) changes in total solar in satellite, ACRIM I on the S program (NOAA-9 and E irradiance increases during minimum. The differences origin (data taken from Le

radiation will differ from reference to Section 1.6. The variation may be due to scattering by atmospheric gases, primarily water vapor. The relationship between the solar constant and the effects may be experimentally determined for a given value of the solar irradiance and a pyranometer reading. A graph of F_{λ} versus T_{λ} at a given wavelength, a graph of F_{λ} versus T_{λ} can be found from a particular measurement of the solar constant T_{λ} can be found from the top of the atmosphere. In the short method, an observation of sky by the position of the sun for 10 to 15 minutes. From the world during a period constant determined by the

variations on the accuracy of the measurements have also been included observations from jet aircraft at about 10 km, VI and VII spacecrafts from these experiments include platforms, a standard of the National Aeronautics

made from satellite by, high-stability satellite- the Nimbus 7 satellite in radiometer. The basic 1.6. Radiometers of this absorbed nearly all the the temperature of the depending to the increase in physical element in a manner al power can be measured

accurately, the temperature response of the radiometer can be calibrated, and it is thus referred to as the *self-calibrating radiometer*.

Solar constant data have been derived from total solar irradiance measurements made by self-calibrating radiometers aboard a number of satellites since 1978. These include the Nimbus 7 Earth Radiation Budget (ERB) mission in 1978; the Solar Maximum Mission (SMM) Active Cavity Radiometer Irradiance Monitor 1 (ACRIM I) in 1980; the Earth Radiation Budget Experiment (ERBE) on board the NASA Earth Radiation Budget Satellite (ERBS, 1984), NOAA 9 (1984), and NOAA 10 (1986); and the ACRIM II measurements on board the Upper Atmosphere Research Satellite (UARS, 1991). Figure 2.12 shows the daily measurements of the solar constant from these satellites from 1979 to 1996. The solid lines are 81-day running means of the daily data. The absolute radiance scale of the ACRIM II data has been adjusted to

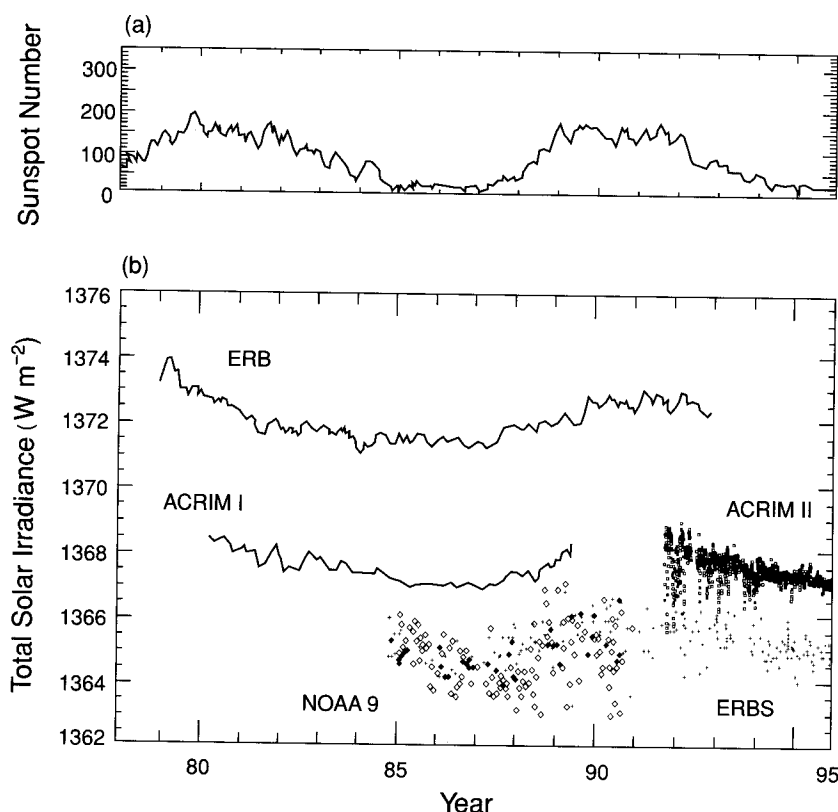


Figure 2.12 Solar activity variations from 1978 to 1996 illustrated by (a) the sunspot number and (b) changes in total solar irradiance. The results were obtained from the ERB radiometer on the Nimbus-7 satellite, ACRIM I on the Solar Maximum Mission (SMM) satellite, ACRIM II on the UARS, and the ERBE program (NOAA-9 and ERBS). The solid lines are 81-day running means of the daily data. Total solar irradiance increases during times of maximum solar activity relative to its levels in the intervening activity minimum. The differences in absolute irradiance levels among various measurements are of instrumental origin (data taken from Lean and Rind, 1998).

match the results of ACRIM I. The differences in absolute irradiance levels among various measurements depicted in this figure are attributed to the instrument sensitivity changes related to temperature or aspect drifts. In particular, ERB and ERBS data differed by about 10 watts per square meter. The top panel shows the sunspot numbers over the same period. It is quite clear that the data displayed in Fig. 2.12 provide irrefutable evidence of the 11-year solar constant cycle. When solar activity is high, as indicated by the sunspot number, the total and UV radiative outputs from the sun increase. Dark sunspots on the solar disk reduce total radiative output because their emission is less than that of the surrounding disk. However, after sunspots develop, magnetic regions involving faculae and plages where emission is enhanced also increase. These regions are evident as complexes of bright emission. The sun's irradiance fluctuates because radiation sources are not homogeneously distributed on its disk. Magnetic fields erupting from the solar convection zone (Section 2.1) into the overlying solar atmosphere generate active regions and complexes in which the local radiation is altered relative to the background solar disk. Magnetic activity erupts, evolves, and decays at different rates throughout the 11-year cycle, generating sunspots, plages, and faculae that modulate total and spectral solar radiative outputs. Finally, it should be noted that our knowledge of the 11-year irradiance cycle is imperfect because of uncertainties arising from the limited duration of space-borne solar monitoring that barely exceeds one 11-year cycle, as well as instrumental uncertainties that cause variable signals in individual satellite solar radiometers.

A number of analyses of the mean total solar irradiance have been reported. Based on the analysis of the solar irradiance measurements taken by the cavity sensor in a number of the satellites depicted in Fig. 2.12, a mean value for the solar constant of 1366 W m^{-2} with a measurement uncertainty of $\pm 3 \text{ W m}^{-2}$ has been suggested (Lean and Rind, 1998). The solar constant value is critical in the interpretation of measured solar absorption and heating rates in the atmosphere.

Exercises

- 2.1 Compute the solar elevation angle at solar noon at the poles, 60° N(S) , 30° N(S) , and the equator. Also compute the length of the day (in terms of hours) at the equator and at 45° N at the equinox and solstice.
- 2.2 From the geometry of an ellipse and the equation defining it, derive Kepler's first law denoted in Eq. (2.2.5).
- 2.3 Based on the conservation of angular momentum that the radius vector drawn from the sun to the planet sweeps out equal areas in equal time, derive Kepler's second law denoted in Eq. (2.2.6).
- 2.4 (a) Derive Kepler's third law by equating Newton's law of universal gravitation and the centrifugal force required to keep the planet in a circular orbit. (b) Given that the NOAA polar satellites orbit at about 850 km above the earth's surface, what would be the period of these satellites? (c) Geostationary satellites have the same angular velocity as the earth. What would be the required height for these satellites?
- 2.5 Given the solar constant 1366 W m^{-2} , the earth-sun distance $150 \times 10^6 \text{ km}$, and the temperature of the sun 5778 K , calculate the solar surface area.
- 2.6 If the average output of the sun is $3.8 \times 10^{26} \text{ W}$, calculate the total energy emitted by the sun in one day?
- 2.7 Compute the fraction of the solar constant that reaches the earth.
- 2.8 Consider a circular disk of radius r at a distance R from the sun. The disk emits radiation toward the earth. Calculate the fraction of the solar constant that reaches the earth over the receiving surface.
- 2.9 Assume that $\bar{\alpha}$ is the average albedo of the earth-atmosphere system. Calculate the amount of flux reflected by the earth-atmosphere system.
- 2.10 The following table gives the albedos of various surfaces. Estimate the temperatures of the surfaces.
- 2.11 The height of earth-orbiting satellites is determined by the orbital radius. GOES satellites, in orbit at $35,786 \text{ km}$ above the earth's surface, calculate the earth-sun distance at the winter solstice, assuming a circular orbit and assuming that the earth's orbit is circular.
- 2.12 Show that the change in the earth-sun distance over the year is small and the sun varies by a small amount. Calculate the change in the earth-sun distance on July 5, respectively, and the temperature.
- 2.13 Calculate the daily variation in the earth-sun distance at the winter solstice. Calculate the earth-sun distance at the winter solstice in Fig. 2.8.

- | | | | | |
|---------------------------|-------|-------|-------|------|
| Zenith angle (degree): | 40° | 50° | 60° | 70° |
| F (W m ⁻²): | 13.95 | 12.55 | 10.46 | 7.67 |

Suggested Reading

- ## Chapter 3

3.1 Composition and

It is now generally accepted that solid materials that condensed from the earth's present atmosphere were generated from volatile elements which the earth formed. The heavy bombardment was lost because the cosmic rays which contains about 90% of the heavy bombardment are extant by 3.5 BY, at which time the post-heavy bombardment with traces of CO and H₂O sphere, associated with the on the earth's surface. A number have evolved to compete suggested that the bioturbation consequence of photosynthesis naturally. The major increase The level of free O₂ is a layer that provided an effect we define the region of

3.1.1 Thermal Struc

To describe the interaction between the atmosphere and the ground, we need to understand the atmospheric conditions in the region of the atmosphere where the solar light, we first present the atmospheric conditions which is shown in Figure 1. The atmospheric conditions in the latitude regions. According to the International Union of Geodesy and Geophysics (IUGG) (1978), the atmosphere is divided into five layers: the troposphere, the stratosphere, the mesosphere, the thermosphere, and the exosphere. The troposphere is the lowest layer of the atmosphere, extending from the surface to an altitude of about 10 km. The stratosphere extends from about 10 km to 50 km. The mesosphere extends from about 50 km to 85 km. The thermosphere extends from about 85 km to 600 km. The exosphere extends from about 600 km to 10,000 km. The atmosphere is composed of various gases, including nitrogen, oxygen, and carbon dioxide. The atmosphere also contains various particles, including dust and water vapor. The atmosphere plays a crucial role in the Earth's climate system, and understanding its conditions is essential for many applications, including climate modeling and satellite remote sensing.

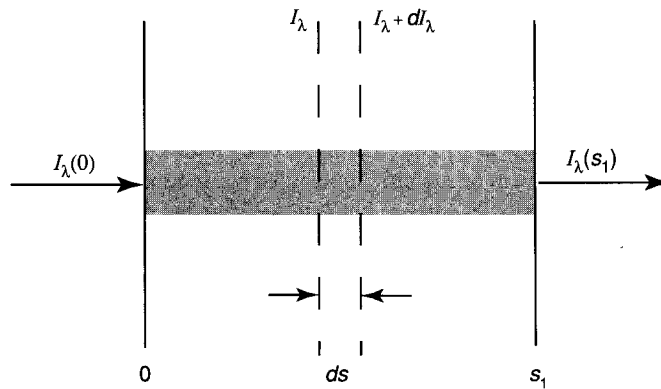


Figure 1.13 Depletion of the radiant intensity in traversing an extinction medium.

In this manner, the source function has units of radiant intensity. It follows that Eq. (1.4.3) may be rearranged to yield

$$\frac{dI_\lambda}{k_\lambda \rho ds} = -I_\lambda + J_\lambda. \quad (1.4.5)$$

This is the general radiative transfer equation without any coordinate system imposed, which is fundamental to the discussion of any radiative transfer process.

1.4.2 Beer–Bouguer–Lambert Law

Consider a direct light beam from the sun, which covers the wavelengths from about 0.2 to 5 μm . Emission contributions from the earth-atmosphere system can be generally neglected, as discussed in Section 1.2. Moreover, if the diffuse radiation produced by multiple scattering can be neglected, then Eq. (1.4.5) reduces to the following form:

$$\frac{dI_\lambda}{k_\lambda \rho ds} = -I_\lambda. \quad (1.4.6)$$

Let the incident intensity at $s = 0$ be $I_\lambda(0)$. Then the emergent intensity at a distance s away shown in Fig. 1.13 can be obtained by integrating Eq. (1.4.6) and is given by

$$I_\lambda(s_1) = I_\lambda(0) \exp\left(-\int_0^{s_1} k_\lambda \rho ds\right). \quad (1.4.7)$$

Assuming that the medium is homogeneous, so that k_λ is independent of the distance s , and defining the path length

$$u = \int_0^{s_1} \rho \, ds, \quad (1.4.8)$$

Eq. (1.4.7) can be expressed by

$$I_\lambda(s_1) = I_\lambda(0)e^{-k_\lambda u}. \quad (1.4.9)$$

This is known as Beer's law. Beer–Bouguer–Lambert law states that the intensity of light traversing a homogeneous medium is a function of the path length. Since this law applies only to the intensity of light,

From Eq. (1.4.9), we obtain

Moreover, for a nonscattered wave, the fractional part of the phase is given by

Equations (1.4.10) and (1.4.11) in conjunction with the assumption that the scattering contribution from the surface will reflect back to the incident wave, yields the reflectivity R_λ , which is the ratio of the reflected intensity to the incident intensity. On the

for the transfer of radiat

1.4.3 Schwarzschild

Consider a nonscattering medium of intensity I_λ passing through it simultaneously. This is the case for the earth and the atmosphere given by the Planck function.

Hence, the equation of

where k_λ is now the absorption coefficient, Eq. (1.4.14) denotes the emission coefficient, the second term represents the emission of the material, and the third term represents the emission of the monochromatic optical radiation.

This is known as Beer's law or Bouguer's law or Lambert's law, referred to here as the Beer-Bouguer-Lambert law, which states that the decrease in the radiant intensity traversing a homogeneous extinction medium is in accord with the simple exponential function whose argument is the product of the mass extinction cross section and the path length. Since this law involves no directional dependence, it is applicable not only to the intensity quantity but also to the flux density and the flux.

From Eq. (1.4.9), we can define the monochromatic transmissivity T_λ as follows:

$$T_\lambda = I_\lambda(s_1)/I_\lambda(0) = e^{-k_\lambda u}. \quad (1.4.10)$$

Moreover, for a nonscattering medium, the monochromatic absorptivity, representing the fractional part of the incident radiation that is absorbed by the medium, is given by

$$A_\lambda = 1 - T_\lambda = 1 - e^{-k_\lambda u}. \quad (1.4.11)$$

Equations (1.4.10) and (1.4.11) are normally expressed in the wavenumber domain in conjunction with the application of infrared radiation transfer. Finally, if there is a scattering contribution from the medium, certain portions of the incident radiation may reflect back to the incident direction. In this case, we may define the monochromatic reflectivity R_λ , which is the ratio of the reflected (backscattered) intensity to the incident intensity. On the basis of the conservation of energy, we must have

$$T_\lambda + A_\lambda + R_\lambda = 1 \quad (1.4.12)$$

for the transfer of radiation through a scattering and absorbing medium.

1.4.3 Schwarzschild's Equation and Its Solution

Consider a nonscattering medium that is in local thermodynamic equilibrium. A beam of intensity I_λ passing through it will undergo the absorption and emission processes simultaneously. This is the case for the transfer of thermal infrared radiation emitted from the earth and the atmosphere. The source function, as defined in Eq. (1.4.4), is given by the Planck function and can be expressed by

$$J_\lambda = B_\lambda(T). \quad (1.4.13)$$

Hence, the equation of radiative transfer can now be written as

$$\frac{dI_\lambda}{k_\lambda \rho ds} = -I_\lambda + B_\lambda(T), \quad (1.4.14)$$

where k_λ is now the absorption coefficient. The first term in the right-hand side of Eq. (1.4.14) denotes the reduction of the radiant intensity due to absorption, whereas the second term represents the increase in the radiant intensity arising from blackbody emission of the material. To seek a solution for Schwarzschild's equation, we define the monochromatic optical thickness of the medium between points s and s_1 as shown